

# Determination of the Maximum $\beta$ -Ray Energy of $Xe^{138}$ (17 min) And $Xe^{137}$ (3.8 min) by Absorption

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The main difficulty in measuring the maximum beta energy of  $Xe^{138}$  is that it can only be determined in the presence of its daughter nuclide,  $Cs^{138}$ , whose disintegration pattern is known and reproduced in Fig. 1, and which has a half-life of 32 minutes with a maximum energy of 2.65 Mev, as determined by absorption.

The same is not true of  $Xe^{137}$ , since its disintegration product,  $Cs^{137}$  (33 years), is not formed in detectable amounts in the time needed for readings to be made on the gas. In this case, it was sufficient to eliminate the interference due to  $Xe^{138}$  and  $Cs^{138}$  with a suitable absorption plate, or to separate it from the  $I^{137}$  previously extracted from the irradiated uranium.

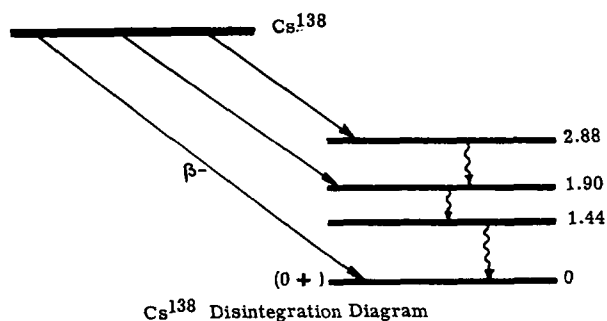


Figure 1

## EXPERIMENTAL

$Xe^{138}$  was obtained by the fission of a uranium solution brought about by thermal neutrons produced by the  $Li(d,n)Be$  reaction in the C. N. E. A. accelerator, with an irradiation time of 15 minutes; also by fission of natural uranium in the synchrocyclotron with 28 Mev deuterons for three minutes, with a current of  $10 \mu a$ .

With the selected irradiation times, the proportion of long lived isotopes is kept very low. Other fission gases, mainly the Kr isotopes, are obtained together with the Xe.

A series of preliminary tests led us to the conclusion that Kr free Xe may be obtained by absorption with activated carbon at a temperature somewhere

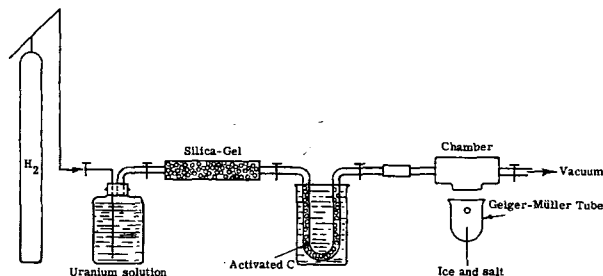


Figure 1'

between  $-12^\circ$  and  $-15^\circ C$ , although the recovery is not quantitative, since some of the Xe is carried away by the Kr.

Therefore, by circulating  $H_2$  for 2 minutes in an apparatus similar to the one illustrated on Fig. 1', the Xe is absorbed in a "U" shaped tube containing activated carbon and submerged in a bath at  $-12^\circ$  to  $-15^\circ C$ .

In order to eliminate the gases not absorbed by the carbon and remaining in the tube, a current of  $H_2$  is circulated. Only the Xe isotopes produced by the fission are obtained under such conditions; the very short lived ones are completely disintegrated during the treatment following irradiation. The 9.6-hour  $Xe^{135}$  and 15-minute  $Xe^{135m}$  isotopes appear in the tube in very small proportions, firstly because their direct formation is very small (Chien-shiung-Wu and Segré<sup>3</sup>), and particularly if the washing with  $H_2$  is done immediately after the radiation, there only being the small amount formed during the operation, from the 6.7-hour  $I^{135}$  produced during the fission.

On the other hand, the  $Xe^{137}$  is almost totally absorbed in the activated carbon, and it is necessary to wait for its complete disintegration.

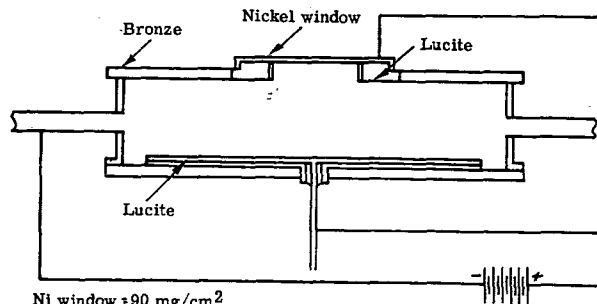


Figure 2

Original language: Spanish.

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The  $Xe^{138}$  was introduced into the gas chamber (Fig. 2) 30 minutes after the end of irradiation; the system including the U-tube and gas chamber having previously been evacuated, and the tube heated in a sulfuric acid bath.

The activity of the  $Xe^{138}$ , plus that of the  $Cs^{138}$  formed by the disintegration of the former, was measured directly as a function of time (Fig. 3).

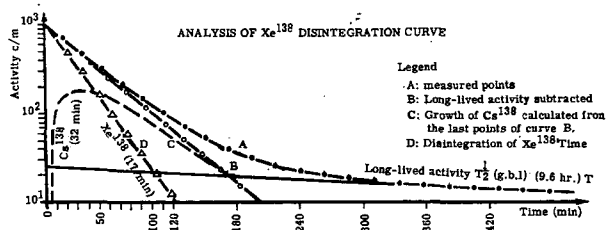


Figure 3

The long-lived activity seen in the curve corresponds to  $Xe^{135}$  (9.6 hours). In the irradiation with deuterons in the synchrocyclotron, we found at  $t = 0$ , 2.5%  $Xe^{135}$  activity relative to  $Xe^{138}$ ; on the other hand, we found only 0.4% when irradiating in the accelerator. In a new experiment we managed to increase the  $Xe^{138}/Cs^{138}$  ratio (Fig. 4) by applying 600 volts to the terminals of a gas chamber so constructed that the  $Cs^{138}$  could be electrically deposited outside the measuring angle (Fig. 2).

This enabled us to construct an absorption curve

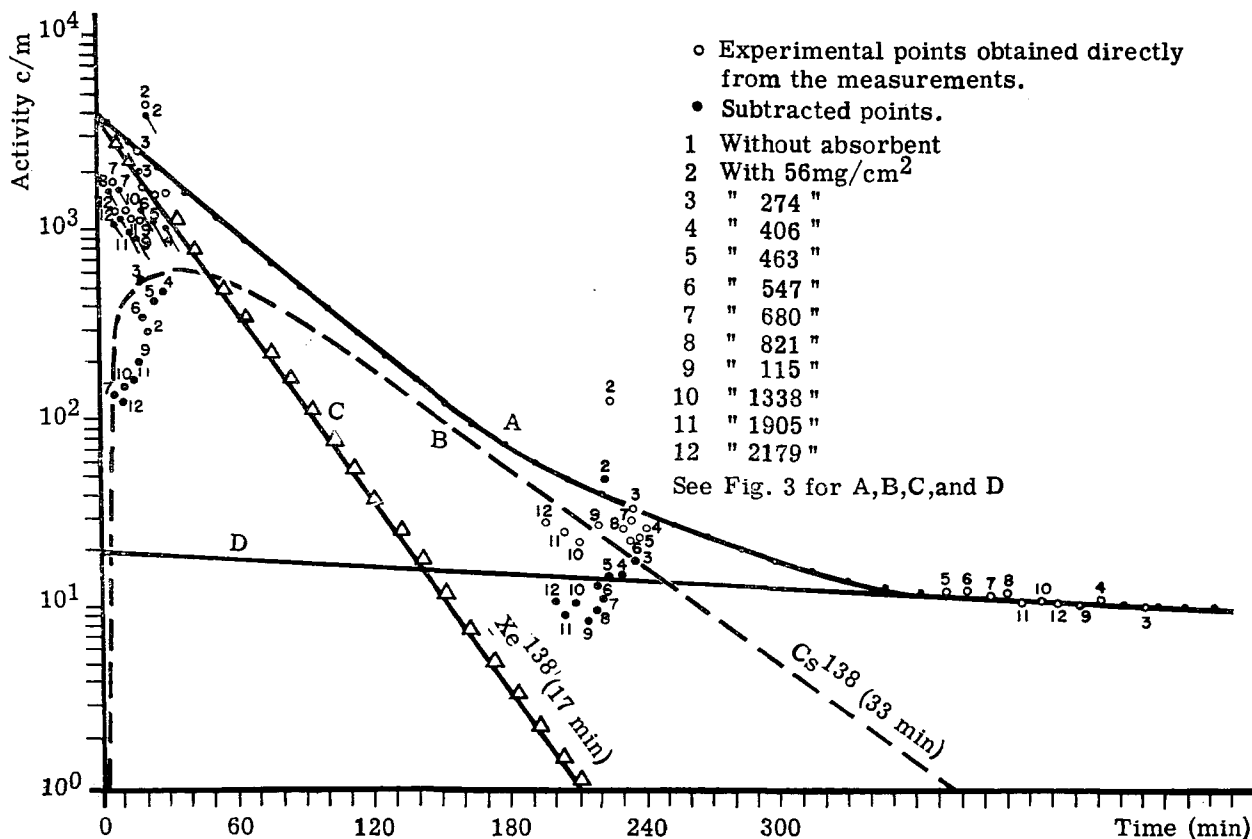


Figure 5

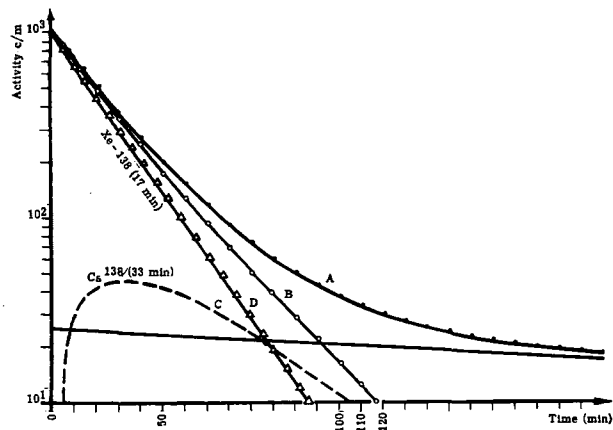


Figure 4. Analysis of the  $Xe^{138}$  disintegration curve obtained by applying 600 volts to the chamber

for  $Xe^{138}$  in which the correction for the influence of  $Cs$  is much smaller.

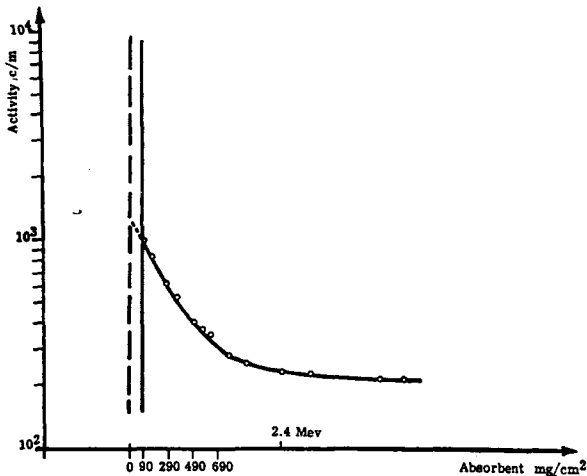
The construction of this curve was as follows:

Figure 5 shows the family of curves, obtained with different thicknesses of aluminum. For clarity, only a few points were marked on it, drawing only the curve corresponding to 274 mg/cm<sup>2</sup>. The numbers indicate the thickness of Al used.

At the points corresponding to  $Cs^{138}$  activity (i.e., after the  $Xe^{138}$  had completely disintegrated), the activity of the  $Xe^{135}$  was subtracted, obtaining in this way the reduction in the activity of the  $Cs^{138}$  for that thickness.

- o Experimental points obtained directly from the measurements.
- Subtracted points.
- 1 Without absorbent
- 2 With 56mg/cm<sup>2</sup>
- 3 " 274 "
- 4 " 406 "
- 5 " 463 "
- 6 " 547 "
- 7 " 680 "
- 8 " 821 "
- 9 " 115 "
- 10 " 1338 "
- 11 " 1905 "
- 12 " 2179 "

See Fig. 3 for A,B,C,and D

Figure 6. Xe<sup>138</sup> absorption curve

From the activities obtained with the same Al thickness in the first minutes when the Xe/Cs is higher (Fig. 4), we subtracted the activity of Cs for that time. The data obtained from different irradiations were normalized to the same value for the initial activity. Figure 6 illustrates the values obtained in this way.

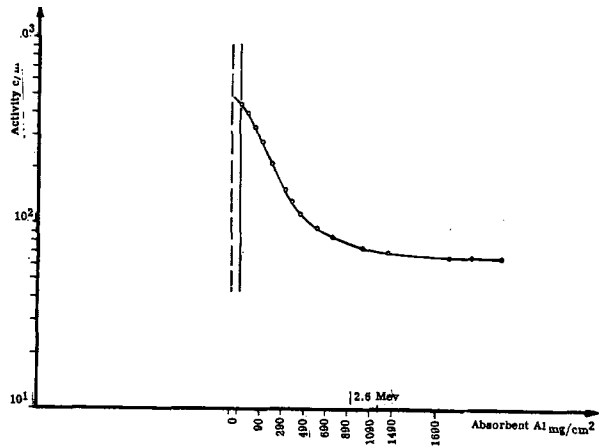
It shows a range of 1135 mg/cm<sup>2</sup>, i.e., an  $E_{max}$  of about 2.4 Mev, if we compare it with the absorption curve for pure Cs<sup>138</sup> (Fig. 7), measured with the same geometry as for the gas. If there are more penetrating  $\beta$ -rays, they must be found in a smaller proportion.

The daughter product was measured in another experiment as follows: after injecting the gas into the chamber, the Cs<sup>138</sup> was allowed to grow until it reached a maximum, after 30 minutes; the Xe<sup>138</sup> was then washed by circulating H<sub>2</sub> and the remaining Cs was measured using different Al thicknesses.

We also found that Xe<sup>138</sup> emits  $\gamma$ -radiations, although we are unable, at this date, to make any statement as to its energy.

We compared our results:  $Q_{\beta} = 2.4 + A$  with the  $Q_{\beta}$  data found in a paper by Way and Wood,<sup>5</sup> and we believe that our experimental value is acceptable, since a discontinuity is observed due to a magic number of neutrons (82) in Xe<sup>136</sup>.

The absorption curve for Xe<sup>137</sup> was measured, separating the gas as was done for Xe<sup>138</sup>, but placing it in the chamber as soon as possible after irradiation. The Al plates used always exceeded the range of the Xe<sup>138</sup> and Cs<sup>138</sup>. A range of about 1790 mg/cm<sup>2</sup> was determined, corresponding to a maximum  $\beta$  energy of 3.5 Mev.

Figure 7. Cs<sup>138</sup> absorption curve

We also checked, with a scintillation counter, that this Xe isotope is a  $\gamma$ -emitter.

### CONCLUSIONS

The maximum beta energies of the isotopes of Xe having mass numbers 138 and 137 were determined by absorption in Al. The values found were as follows: for Xe<sup>138</sup>: 2.4 Mev; for Xe<sup>137</sup>: 3.5 Mev.

It was very useful, in the first case, to enclose the gas in a special chamber and to apply a potential difference in order to have the Cs electrically deposited outside of the angle of measurement.

$\gamma$ -radiation was detected in the two Xe isotopes.

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