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## Interconfigurational fluctuations in Ce cubic compounds

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**Résumé.** — La susceptibilité magnétique de trente composés cubiques de cérium est analysée, en supposant une dépendance en température des fluctuations de valence. Le schéma des niveaux d'énergie obtenu par l'ajustement de la susceptibilité mesurée est utilisé pour calculer d'autres propriétés physiques. On a trouvé une corrélation entre le rayon ionique du cérium et l'énergie d'excitation interconfigurationnelle.

**Abstract.** — The magnetic susceptibility of thirty Ce cubic compounds is analyzed, assuming a temperature dependence of the valence fluctuations. The energy level scheme, obtained by fitting the measured susceptibility is used to calculate other physical properties. A correlation between the Ce ionic radius and the interconfigurational excitation energy has been found.

The magnetic susceptibility (MS) of thirty Ce cubic intermetallic compounds (IC), of different crystal structures (see the table I), which shows magnetic, intermediate and non magnetic behaviours, has been analyzed assuming a temperature dependence of the valence fluctuations. We have applied the concept of interconfigurational fluctuations (ICF) [1, 2] to describe the observed temperature dependence of the MS [3], taking into account the possibility of a dynamical mixing between the two lower energy Ce configurations ( $4f^1 5d^1 \rightleftharpoons 4f^0 5d^2$ ), identified as  $n = 1$  and  $n = 0$  respectively. From this model we have derived a sufficiently accurate energy level scheme to attempt to fit available experimental data of different physical properties.

The probability of occupation of the state «  $n$  » (of energy  $E_n$  and width  $\Delta_n$ ), is given at  $T \rightarrow 0$  by :  $P_n(0) \sim \exp(-E_n/\Delta_n)$ , and at high temperatures by :  $P_n(T) \sim \exp(-E_n/k_b T)$ . A good interpolation is done considering  $P_n(T) \sim \exp(-E_n/T_n)$  [4], where  $T_n = T + \Delta_n$ , and  $k_b = 1$  using K as energy unit. So, the valence dependence with temperature was taken as the *fractional occupation* of the configuration  $4f^1 5d^1$  ( $n = 1$ ), i.e.

$$V(T) = 1/Z \int_{-\infty}^{\infty} D_n(E) f_n^x(E) dE,$$

where  $f_n^x(E) = \exp(-E_n/T_n)$  is the quasi Boltzmann distribution and «  $Z$  » the partition function,

$$Z = \sum_n (2J_n + 1) f_n^x(E).$$

Taking  $D_n(E) \sim \exp[-(E - E_n)^2/2 \Delta_n^2]$  it is possible to integrate, obtaining :

$$V(T) = (2J_n + 1) \exp[\frac{1}{2}(\Delta_n/T_n)^2 - E_n/T_n]^{1/2}.$$

Substituting  $T$  by  $T_n$ , the MS can be written as  $\chi(T) = (N/3 T_n) \sum \mu_n^2 V(T)$ . In the case of Ce it is :  $\mu_1 = 2.54 \mu_B$  and  $\mu_0 = 0$ . To apply this expression to a real case, it is necessary first to account for other effects, which will be considered as independent from ICF. They are : 1) The possible interactions between magnetic moments, which give rise to a Curie-Weiss temperature ( $T_0$ ), 2) the Van Vleck contribution to the MS, independent from temperature, and 3) the possible splitting, into a doublet  $\Gamma_7$  and a quartet  $\Gamma_8$ , of the Hund sixfold ground state due to the crystal field effect. These exchange interactions will be represented by  $|\langle \Gamma_i | J_z | \Gamma_j \rangle| = J_{ij}$ . Under these considerations the general expression for the MS becomes :

$$\chi(T) = N g_n^2 \mu_B^2 / Z \left\{ \left[ \sum_{i,j} \left( \frac{J_{ii}^2}{T_n - T_0} + \frac{2 J_{ij}^2}{E_j - E_i} \right) \exp(-E_n/T_n) \right] + \frac{2 J_{m'}^2}{E' - E_n} \exp(-E'/T) \right\}$$

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Table I.

Compound	Structure type	Lattice parameter (Å)	Ce ionic radius (Å)	$E_{ex}$ (K)	Magnetic behaviour	Specific heat calc. meas. (mJ/K mole)		
CeSn <sub>3</sub>	AuCu <sub>3</sub>	4.722	1.013	190	Intermed.	64	53	
CePd <sub>3</sub>		4.163	1.009	360		29	36	
CeRh <sub>3</sub>		4.025	0.980	1 600				
CeTl <sub>3</sub>		4.771	1.034	850	Magnetic			
CeIn <sub>3</sub>		4.691	1.034	$E_{ex} > E'$				
CePb <sub>3</sub>		4.875	1.025	550				
Ce <sub>3</sub> Al		5.013	1.020	$E_{ex} > E'$	Magnetic			
Ce <sub>3</sub> Ga		5.115						
Ce <sub>3</sub> In		5.061	1.040					
CeNi <sub>2</sub>	MgCu <sub>2</sub>	7.208	0.954	470	Non mag.	10.3	25.6	
CeRu <sub>2</sub>		7.543	0.920	1 260		33	30	
CeRh <sub>2</sub>		7.546	0.977	420				
CeOs <sub>2</sub>		7.593	0.938	430				
CeIr <sub>2</sub>		7.576	0.964	910				
CePt <sub>2</sub>		7.741	1.034		Magnetic			
CeAl <sub>2</sub>		8.060	1.030	$E_{ex} > E'$	Magnetic			
CeN	NaCl	5.022	0.955	2 150	Non mag.	0	8	
CeP		5.913	1.020	520	Magnetic			
CeAs		6.086	1.037	1 160				
CeSb		6.429	1.032	700				
CeBi		6.505	1.031	860				
CeAg			3.736	1.032	$E_{ex} > E'$	Magnetic		
CeMg		3.924	1.034					
CeZn	CsCl	3.713	1.030					
CeHg		3.808	1.030					
CeTi		3.893	1.033					
CeCd <sub>3</sub>		BiLi <sub>3</sub>	7.218				$E_{ex} > E'$	Magnetic
CeBe <sub>13</sub>	NaZn <sub>13</sub>	10.375	1.020	320	Intermed.	95	115	
CeB <sub>6</sub>	CaB <sub>6</sub>	4.140	1.022	80	Intermed.			
Ce <sub>2</sub> C <sub>3</sub>	Pu <sub>2</sub> C <sub>3</sub>	8.447	0.952	1 200	Non mag.			

where  $g_n = 6/7$  and  $E'$  is the energy of the first  $J$ -excited level characterized by  $J' = 7/2$ . The level scheme was obtained by fitting this formula to the actual experimental data [3].

In table I we give the values of the excitation energy between configurations :  $E_{ex} = E_{n-1} - E_n$ , obtained from this method, as well as the comparison between the calculated and the measured *specific heat* values at  $T = 1$  K for some compounds. The calculated *thermoelectric power* of CeSn<sub>3</sub> and CeBe<sub>13</sub> [5], *nuclear*

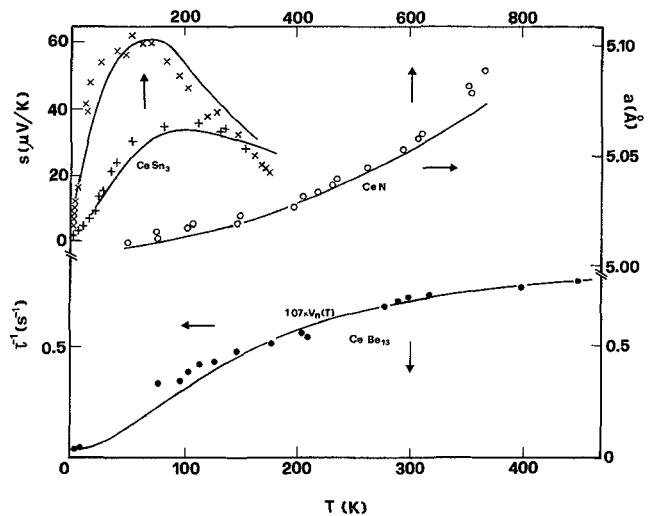


Fig. 1. — Experimental data and theoretical fitting curve of the temperature dependence of : CeBe<sub>13</sub> (x) and CeSn<sub>3</sub> (+) thermoelectrical power (upper right scale), CeN (o) lattice parameter (upper left scale) and CeBe<sub>13</sub> (●) nuclear spin-lattice relaxation rate (lower scale).

spin-lattice relaxation rate of  $\text{CeBe}_{13}$  [6] and the lattice parameter of  $\text{CeN}^7$  as a function of temperature are compared with the experimental data in figure 1.

As it is well known there exists a strong dependence of the Ce atomic volume from the valence. To confirm the fact that all possible behaviours of Ce in intermetallic compounds can be described from its two lower energy configurations, we have plotted in figure 2 the Ce ionic radius ( $r_i$ ) [3] vs.  $E_{\text{ex}}$  of the compounds where  $E_{\text{ex}} < E'$ . It is remarkable that the compounds whose structures are :  $\text{AuCu}_3$ ,  $\text{Pu}_2\text{C}_3$ ,  $\text{NaCl}$ ,  $\text{NaZn}_{13}$  and  $\text{CaB}_6$  are practically on the same line. The departure from this general behaviour of the Laves phases ( $\text{MgCu}_2 \equiv \text{AB}_2$ ) can be understood taking into account the extra compression supported by the element A to obtain the ratio  $r_A/r_B \approx 1.33$ , which characterises these phases.

A more general expression of  $V(T)$  is actually studied using the Fermi distribution for  $f_n(E)$  and a Lorentzian function for  $D_n(E)$  in view of having a better fit also in the case of  $E_n \ll T$ .

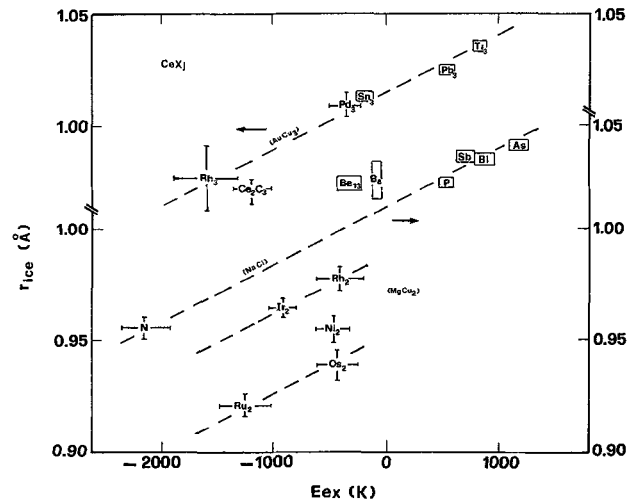


Fig. 2. — Ce ionic radius ( $r_{\text{ice}}$ ) versus the interconfigurational excitation energy ( $E_{\text{ex}}$ ) of the compounds having  $E_{\text{ex}} < E'$ . Only the Ce partner is marked (except in the case of  $\text{Ce}_2\text{C}_3$ ).

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